

X-ray-induced electronic damage of matter by post-ionization mechanisms

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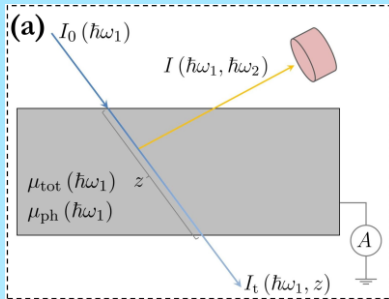


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Introduction

Radiation damage

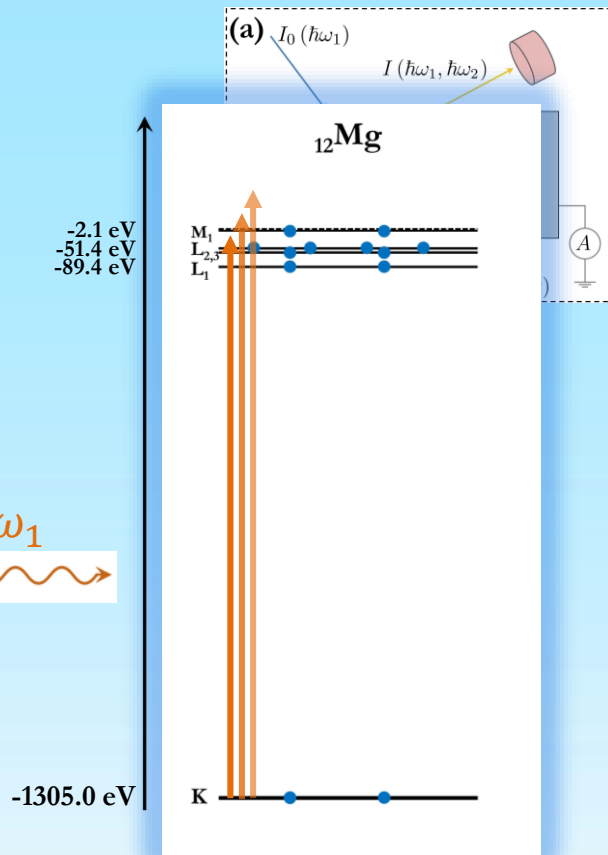
X-ray spectroscopy (XAS, XES)



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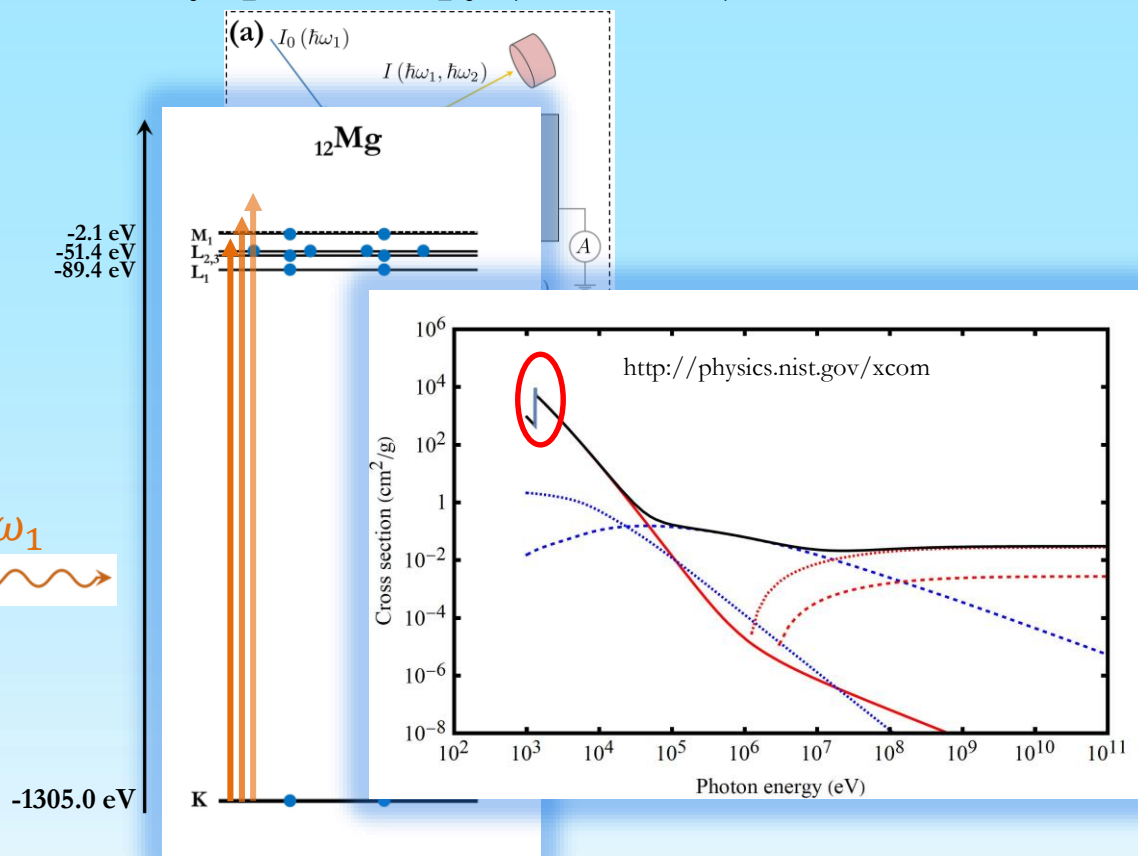
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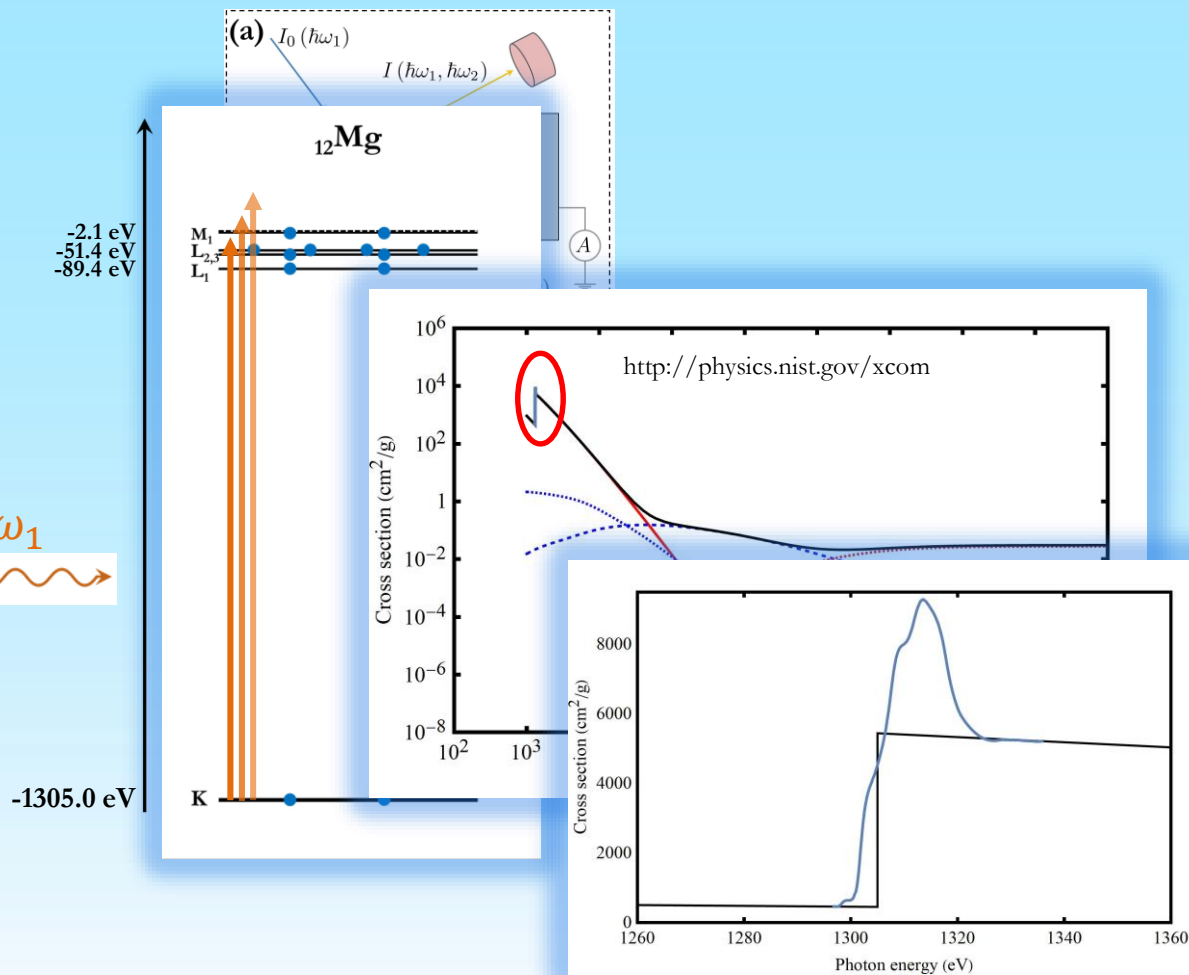
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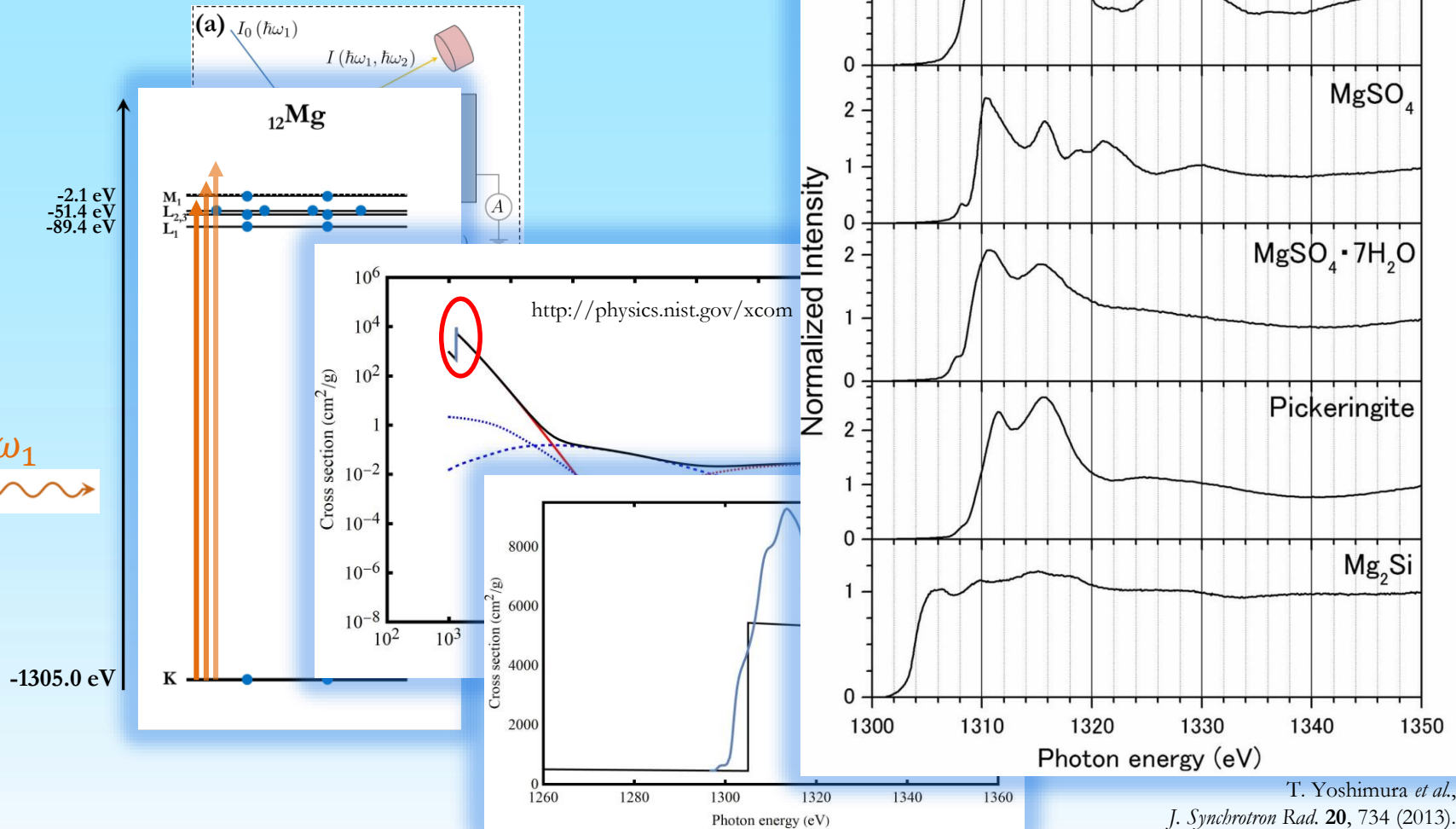
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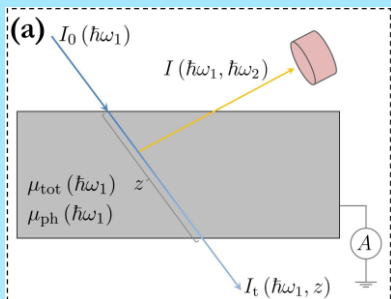
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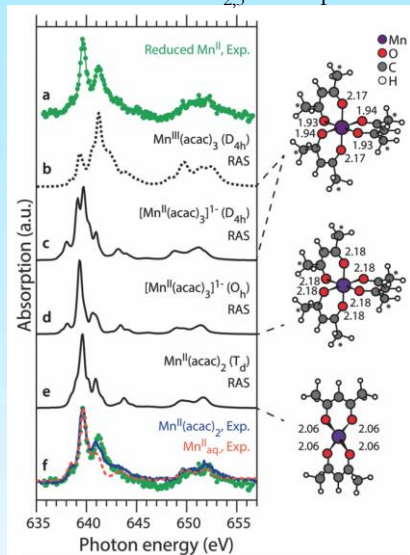
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Mn $L_{2,3}$ XAS spectrum



M. Kubin *et al.*,
Phys. Chem. Chem. Phys. **20**, 1681 (2018).



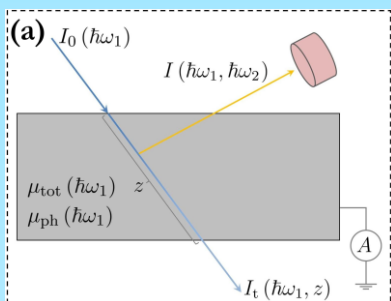
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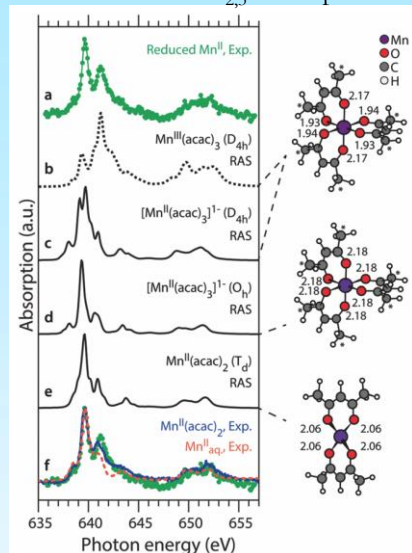
Radiation damage

Radiation damage manifests itself through different effects like electronic structure change, bond breaking, Coulomb explosion, and structural changes.

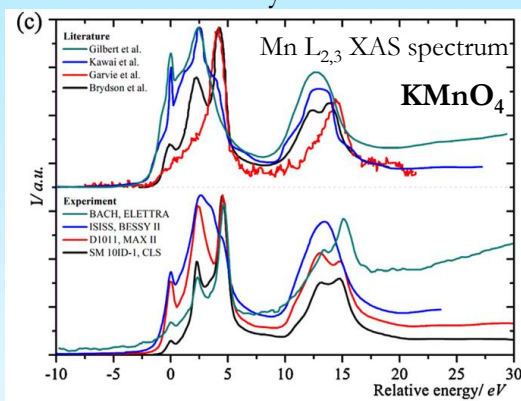
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Mn L_{2,3} XAS spectrum



Systematic review of spectral differences in literature data in a function of X-ray dose.



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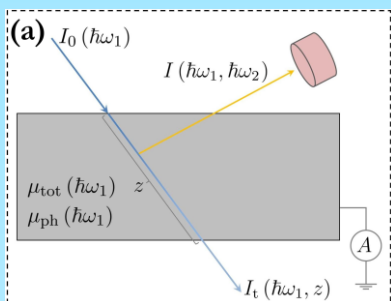
J. El. Spectr. Rel. Phenom. **198**, 31–56 (2015) EWODNICZANSKI
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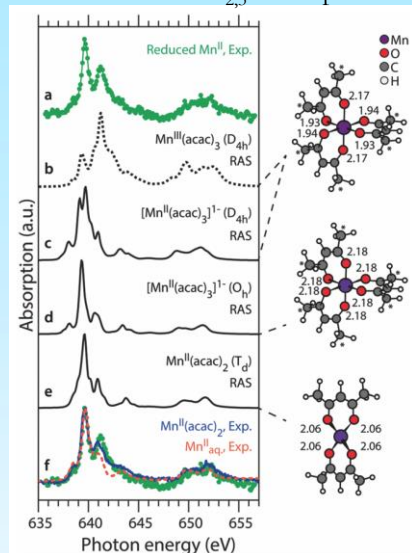
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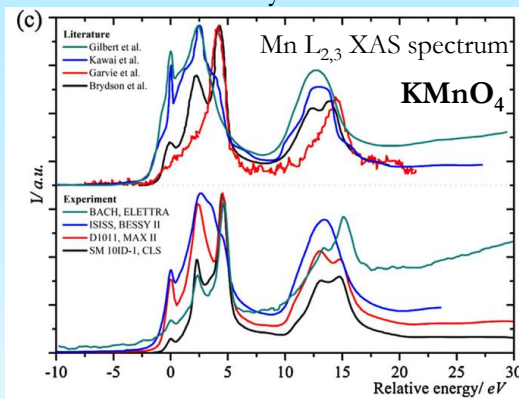
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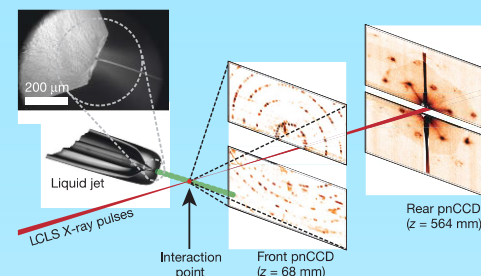
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X-ray diffraction (XRD)



H. N. Chapman *et al.*, *Nature* **470**, 73–77 (2011).

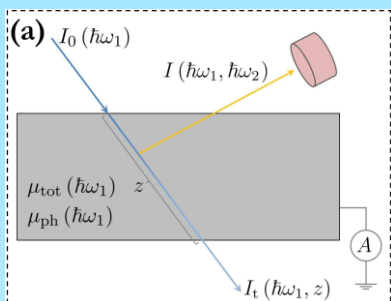
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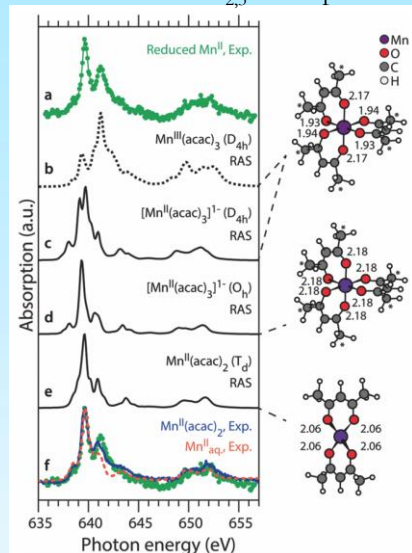
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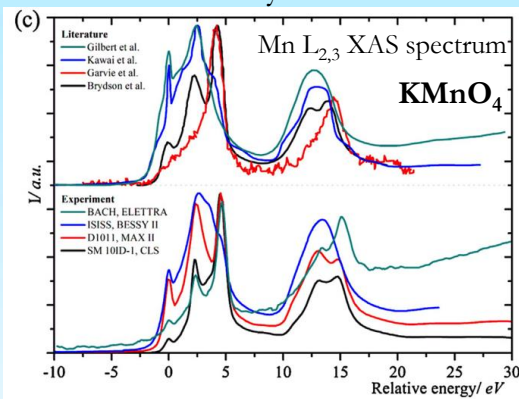
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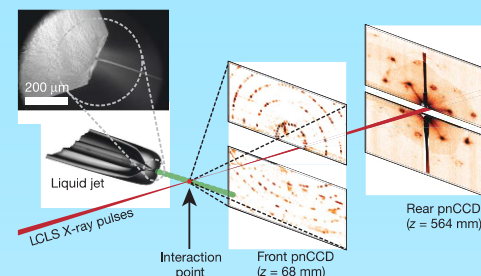
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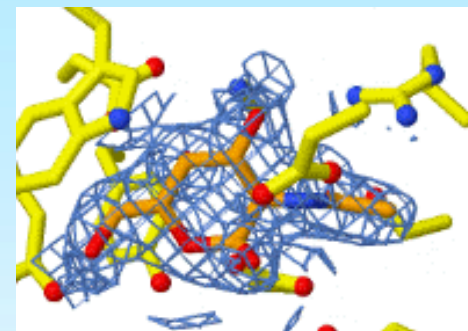
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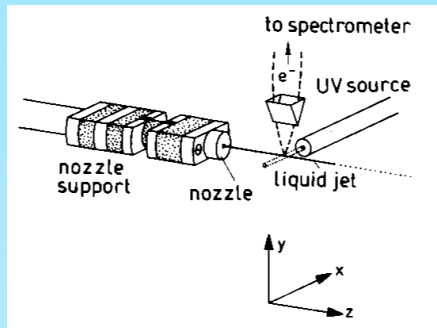
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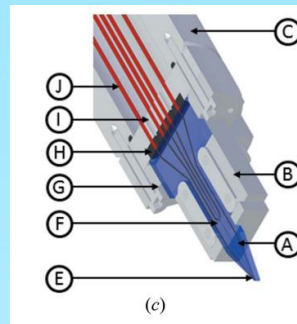
Introduction

Radiation damage - prevention

Liquid sample delivery systems



M. Faubel, B. Steiner, J. P. Toennies,
J. Chem. Phys. **106**, 9013 (1997).



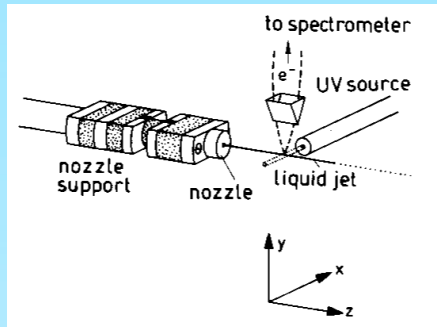
J. Schulz *et al.*,
J. Synchrotron Rad. **26**, 339
(2019).



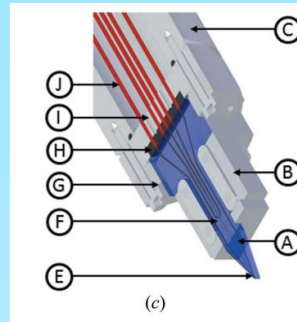
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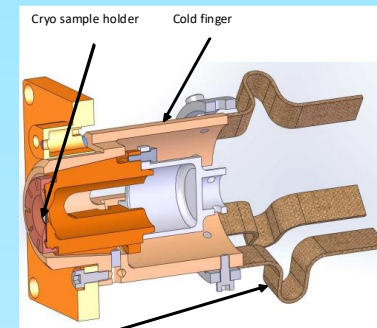


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Cryo-cooling techniques



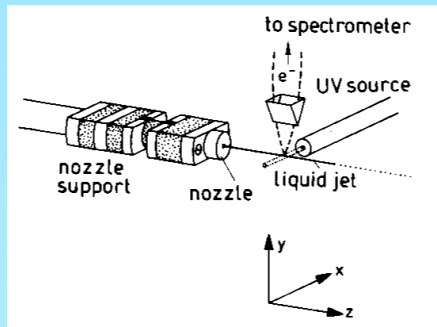
M. Salome *et al.*, *J. Phys* **425**,
182004 (2013).



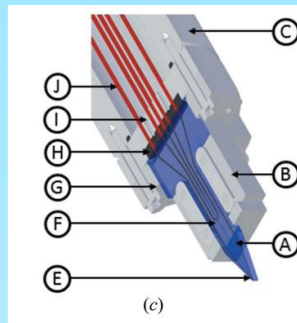
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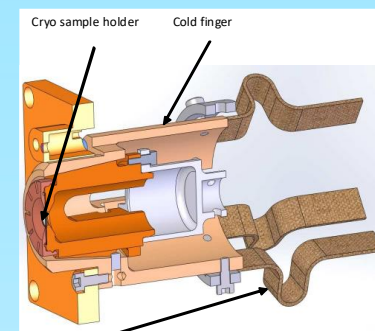


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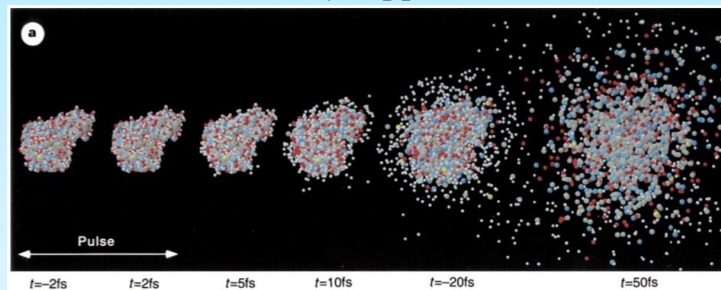
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(2019).

Cryo-cooling techniques



M. Salome *et al.*, *J. Phys* **425**,
182004 (2013).

“Probe-before-destroy” approach



R. Neutze *et al.*, *Nature* **406**, 752 (2000).



Introduction

X-ray free-electron laser (XFEL) beam interaction with matter

Nature **466**, 56 (2010)

Femtosecond electronic response of atoms to ultra-intense X-rays

L. Young¹, E. P. Kanter¹, B. Krässig¹, Y. Li¹, A. M. March¹, S. T. Pratt¹, R. Santra^{1,2}, S. H. Southworth¹, N. Rohringer³, L. F. DiMauro⁴, G. Doumy⁴, C. A. Roedig⁴, N. Berrah⁵, L. Fang⁵, M. Hoener^{5,6}, P. H. Bucksbaum⁷, J. P. Cryan⁷, S. Ghimire⁷, J. M. Glowina⁷, D. A. Reis⁷, J. D. Bozek⁸, C. Bostedt⁸ & M. Messerschmidt⁸

[...] At this fluence, the neon target inevitably changes during the course of a single femtosecond-duration X-ray pulse—by sequentially ejecting electrons—to produce fully-stripped neon through absorption of six photons. Rapid photoejection of inner-shell electrons produces ‘hollow’ atoms and an intensity-induced X-ray transparency. Such transparency, due to the presence of inner-shell vacancies, can be induced in all atomic, molecular and condensed matter systems [...]



beam parameters:

2000 eV (above Ne *K* edge)

pulse duration: 40 fs

$10^{11} - 10^{12}$ photons per pulse

beam spot size: $1.0 \times 1.0 \mu\text{m}^2$

Science **340**, 491 (2013)

Simultaneous Femtosecond X-ray Spectroscopy and Diffraction of Photosystem II at Room Temperature

Jan Kern^{1,2}, Roberto Alonso-Mori², Rosalie Tran¹, Johan Hattne¹, Richard J. Gildea¹, Nathaniel Echols¹, Carina Glöckner³, Julia Hellmich³, Hartawan Laksmono⁴, Raymond G. Sierra⁴, Benedikt Lassalle-Kaiser^{1,‡}, Sergey Koroidov⁵, Alyssa Lampe¹, Guangye Han¹, Sheraz Gul¹, Dörte DiFiore³, Despina Milathianaki², Alan R. Fry², Alan Miahnahri², Donald W. Schafer², Marc Messerschmidt², M. Marvin Seibert², Jason E. Koglin², Dimosthenis Sokaras⁶, Tsu-Chien Weng⁶, Jonas Sellberg^{6,7}, Matthew J. Latimer⁶, Ralf W. Grosse-Kunstleve¹, Petrus H. Zwart¹, William E. White², Pieter Glatzel⁸, Paul D. Adams¹, Michael J. Bogan^{2,4}, Garth J. Williams², Sébastien Boutet², Johannes Messinger⁵, Athina Zouni³, Nicholas K. Sauter¹, Vittal K. Yachandra^{1,*}, Uwe Bergmann^{2,*}, and Junko Yano^{1,*}

[...] Our simultaneous XRD/XES study shows that the PS II crystals are intact during our measurements at the LCLS, not only with respect to the structure of PS II, but also with regard to the electronic structure of the highly radiation sensitive Mn_4CaO_5 cluster, opening new directions for future dynamics studies.



beam parameters:

7000 eV (above Mn *K* edge)

pulse duration: 45 fs

$3 - 6 \times 10^{11}$ photons per pulse

beam spot size: $1.5 \times 1.5 \mu\text{m}^2$



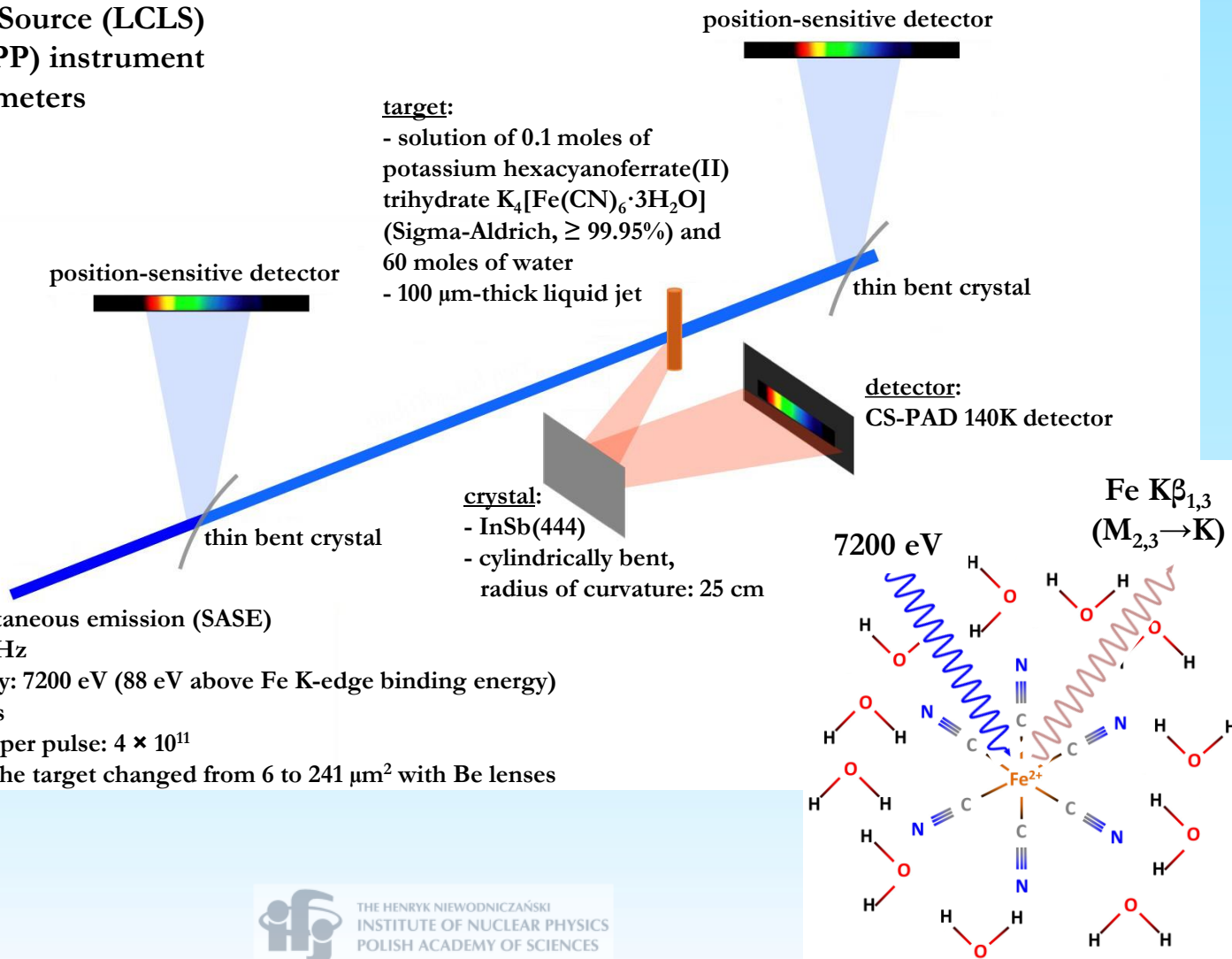
Experimental

Fe $K\beta$ X-ray emission spectroscopy (XES) at X-ray free-electron laser (XFEL) for changing photon flux

Linac Coherent Light Source (LCLS) X-ray pump-probe (XPP) instrument typical operation parameters

X-ray beam:

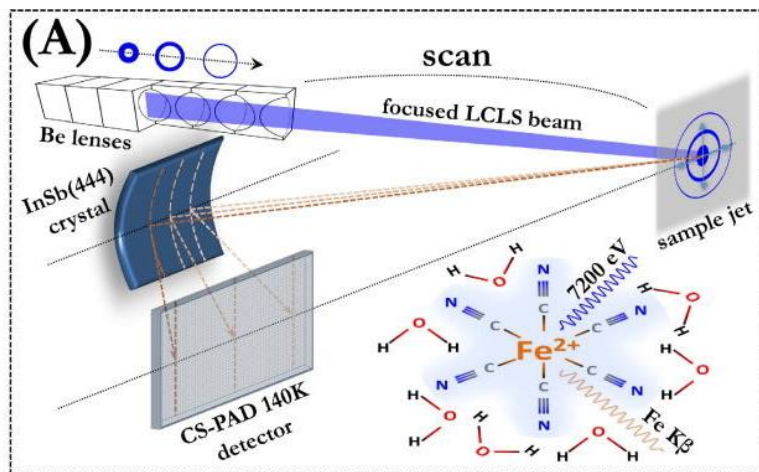
- self-amplified spontaneous emission (SASE)
- repetition rate: 120 Hz
- mean photon energy: 7200 eV (88 eV above Fe K-edge binding energy)
- pulse duration: 30 fs
- number of photons per pulse: 4×10^{11}
- beam spot area on the target changed from 6 to $241 \mu\text{m}^2$ with Be lenses



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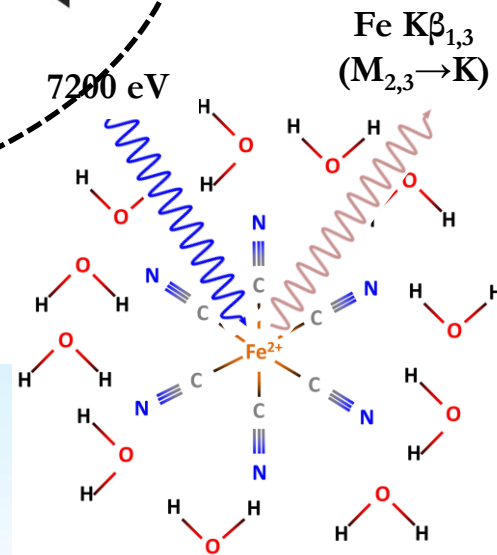
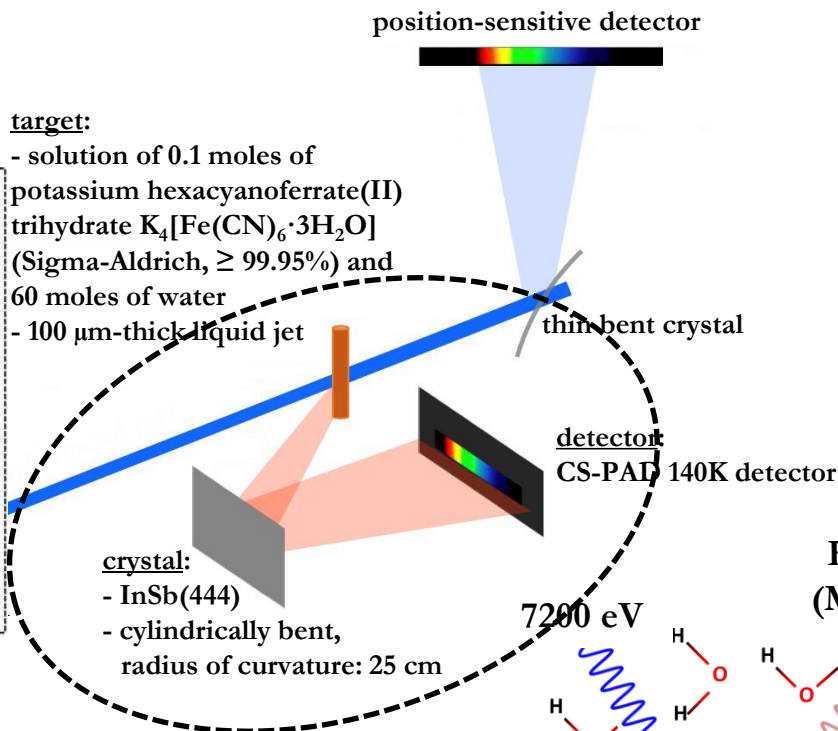
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target:

- solution of 0.1 moles of potassium hexacyanoferrate(II) trihydrate $\text{K}_4[\text{Fe}(\text{CN})_6 \cdot 3\text{H}_2\text{O}]$ (Sigma-Aldrich, $\geq 99.95\%$) and 60 moles of water
- 100 μm -thick liquid jet

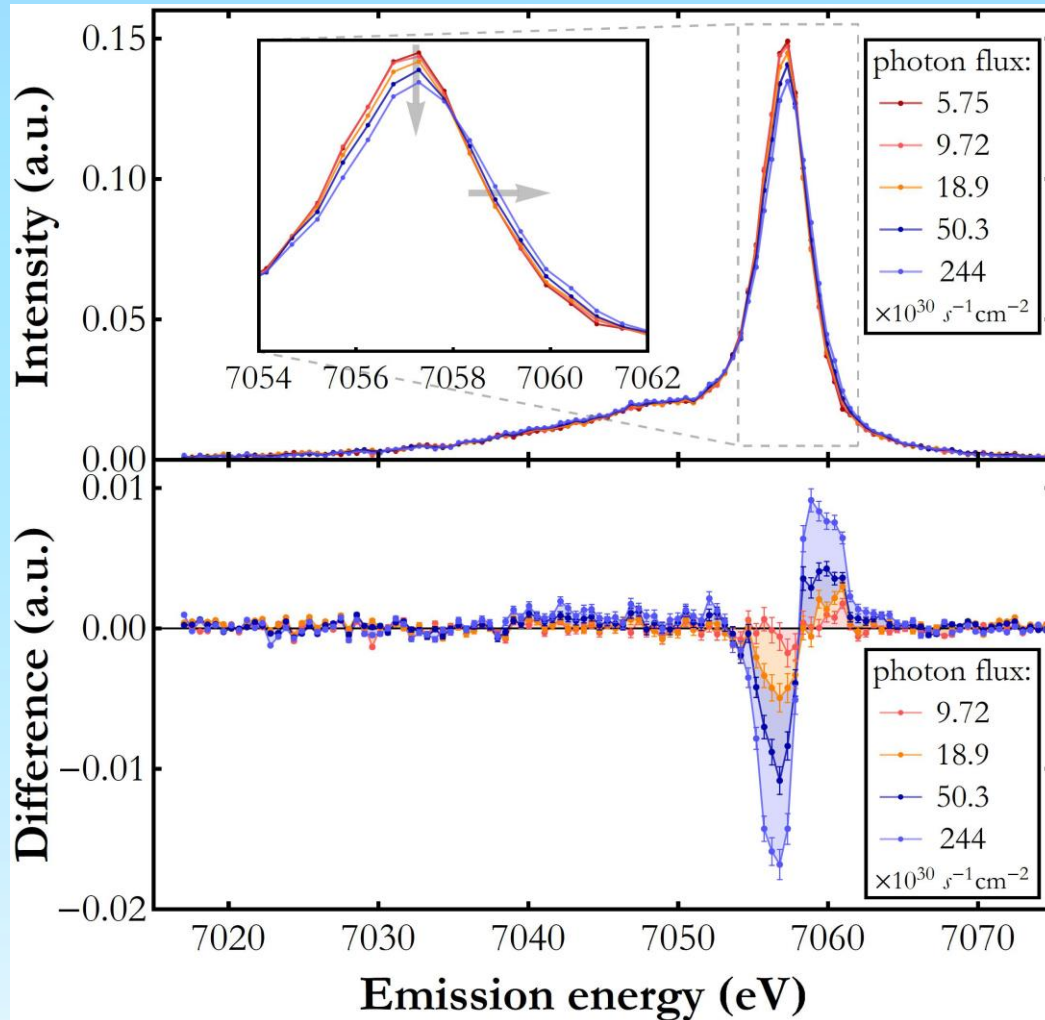
crystal:

- InSb(444)
- cylindrically bent, radius of curvature: 25 cm



Results and discussion

Fe K β dependence on the incident photon flux



The Fe K β X-ray emission spectra measured for $\text{K}_4[\text{Fe}(\text{CN})_6 \cdot 3\text{H}_2\text{O}]$ water solution for different fluxes of the incident photon beam.

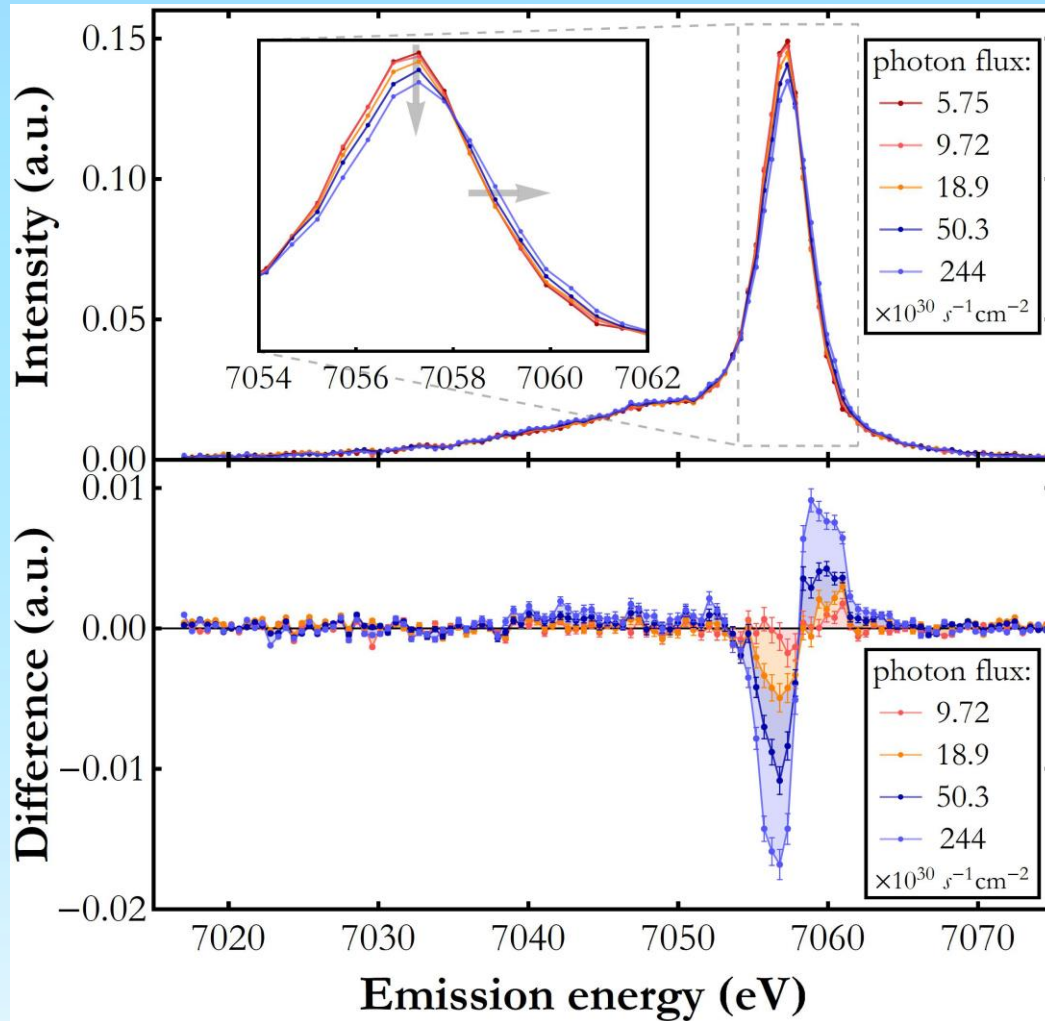
The spectra were scaled to obtain the integral over entire studied energy domain equal to 1 to ensure constant $1s$ shell ionization cross section and the $3p \rightarrow 1s$ transition fluorescence yield and is supported by experimental studies showing that the influence of oxidation state change on the ratio of Fe K α to K β fluorescence yields is below 10 %. The intensity measurement uncertainty does not exceed 2 % in the energy range 7054 – 7062 eV.

S. K. Kulshreshtha, D. N. Wagh, H. N. Bajpei, *X-Ray Spectrom.* **34**, 200-202 (2005).

O. K. Köksal, G. Apaydın, E. Cengiz, K. Karabulut, *J. Phys.: Conf. Ser.* **707**, 012004 (2016).

Results and discussion

Fe $K\beta$ dependence on the incident photon flux



- not caused by the molecule's spin change as in W. Zhang *et al.*, *Nature* **509**, 345 (2014).

- not caused by sequential photoionization as, e.g., in A. Rudenko *et al.*, *Nature* **546**, 129 (2017).



Understanding the mechanism

Motivation

15 % of the incident photons are absorbed, 82 % of the photoabsorption events result in O 1s photoionization.

T. Schoonjans *et al.*, *Spectrochim. Acta, Part B* **66**, 776-784 (2011).



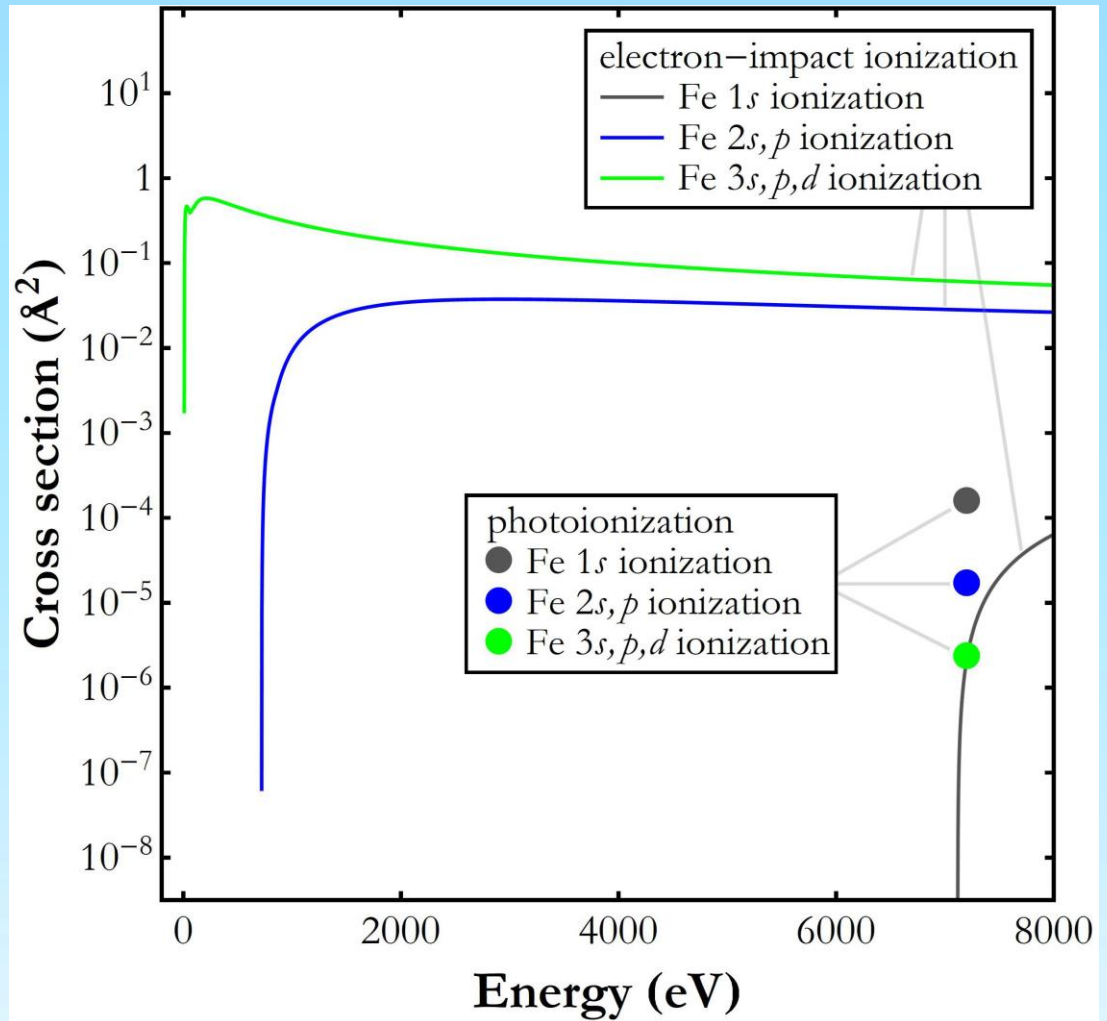
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The cross section for electron-impact Fe ionization is up to the order of 10^4 larger than the Fe 1s shell photoionization cross section.



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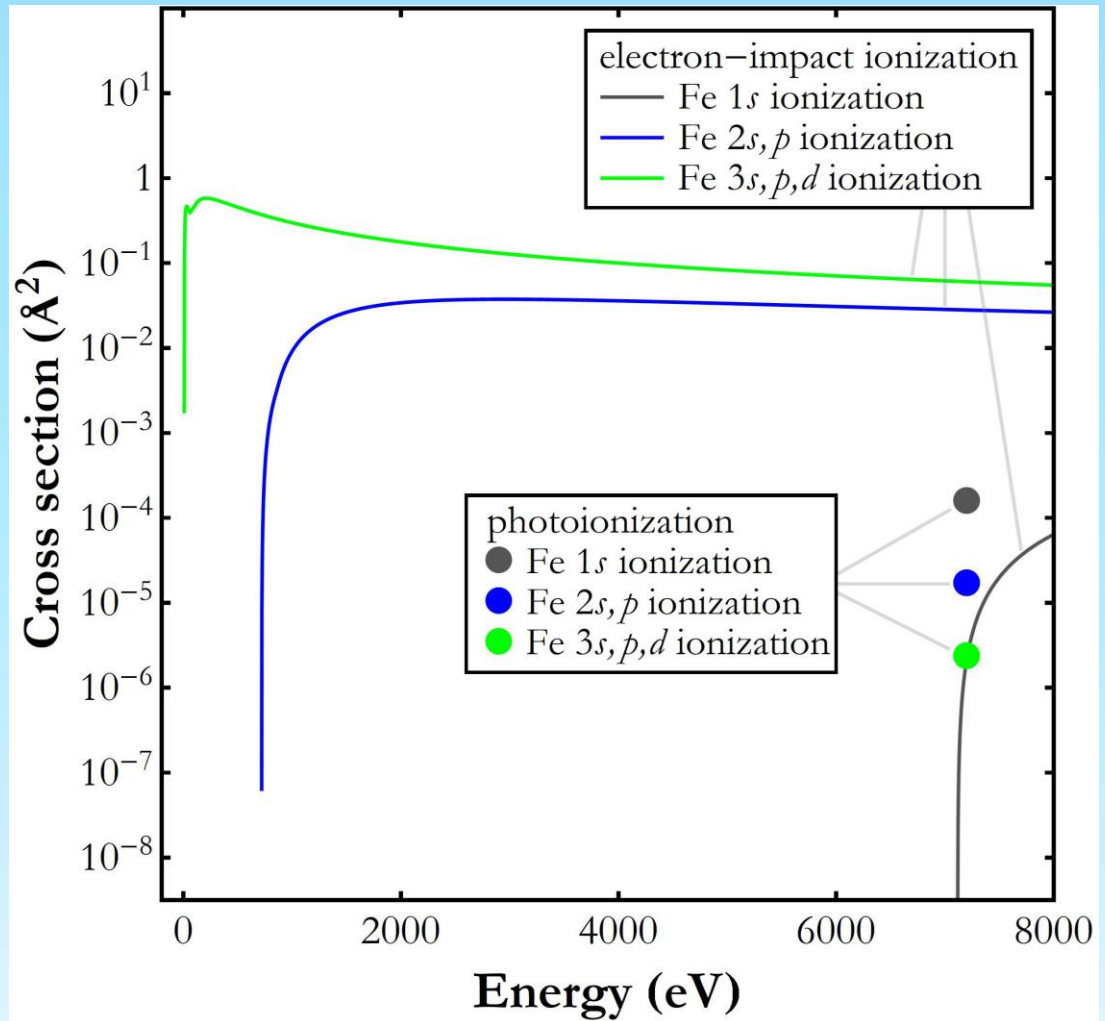
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Photoelectrons and Auger electrons travel in average tens of nm between interactions and may reach the nearest Fe atoms in less than 1 fs.

Unlike photons, free electrons interact with the material multiple times losing the energy gradually.



T. Schoonjans *et al.*, *Spectrochim. Acta, Part B* **66**, 776-784 (2011).

B. Tsipinyuk, A. Bekkerman, E. Kolodney, <https://arxiv.org/abs/physics/0407030> (2005).

H. Shinotsuka *et al.*, *Surf. Interface Anal.* **49**, 238-252 (2016).

Understanding the mechanism

Energy and time distribution of photons and electrons, Fe ionization probability

$$dE = -\overline{E_{loss,e\leftrightarrow a}}(E) f_e(E) dt$$

E - electron energy, t - time

$f_e(E)$ - average collision frequency
per electron

$\overline{E_{loss,e\leftrightarrow a}}(E)$ - average energy loss
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P. M. Banks, *Planet. Space Sci.* **14**, 1085-1103 (1966).

V. Vinogradov, V. P. Shevel'ko, *Sov. Phys. JETP* **44**, 542 (1976).

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$$f_e(E) = n_a\sigma_{i,e\leftrightarrow a}(E)\bar{v}_e(E)$$

n_a - concentration of atoms a

$\sigma_{i,e\leftrightarrow a}(E)$ - cross section for
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$$S_a(E) = n_a \sigma_{i,e\leftrightarrow a}(E) \overline{E_{loss,e\leftrightarrow a}}(E)$$

$S_a(E)$ - linear stopping power for a

P. M. Banks, *Planet. Space Sci.* **14**, 1085-1103 (1966).

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J. Appl. Phys. **103**, 063707 (2008).

Phys. Med. Biol. **30**, 331-335 (1985).



Understanding the mechanism

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$$dE = -\overline{E_{loss,e\leftrightarrow a}}(E)f_e(E)dt$$

E - electron energy, t - time

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$\overline{E_{loss,e\leftrightarrow a}}(E)$ - average energy loss per interaction

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n_a - concentration of atoms a

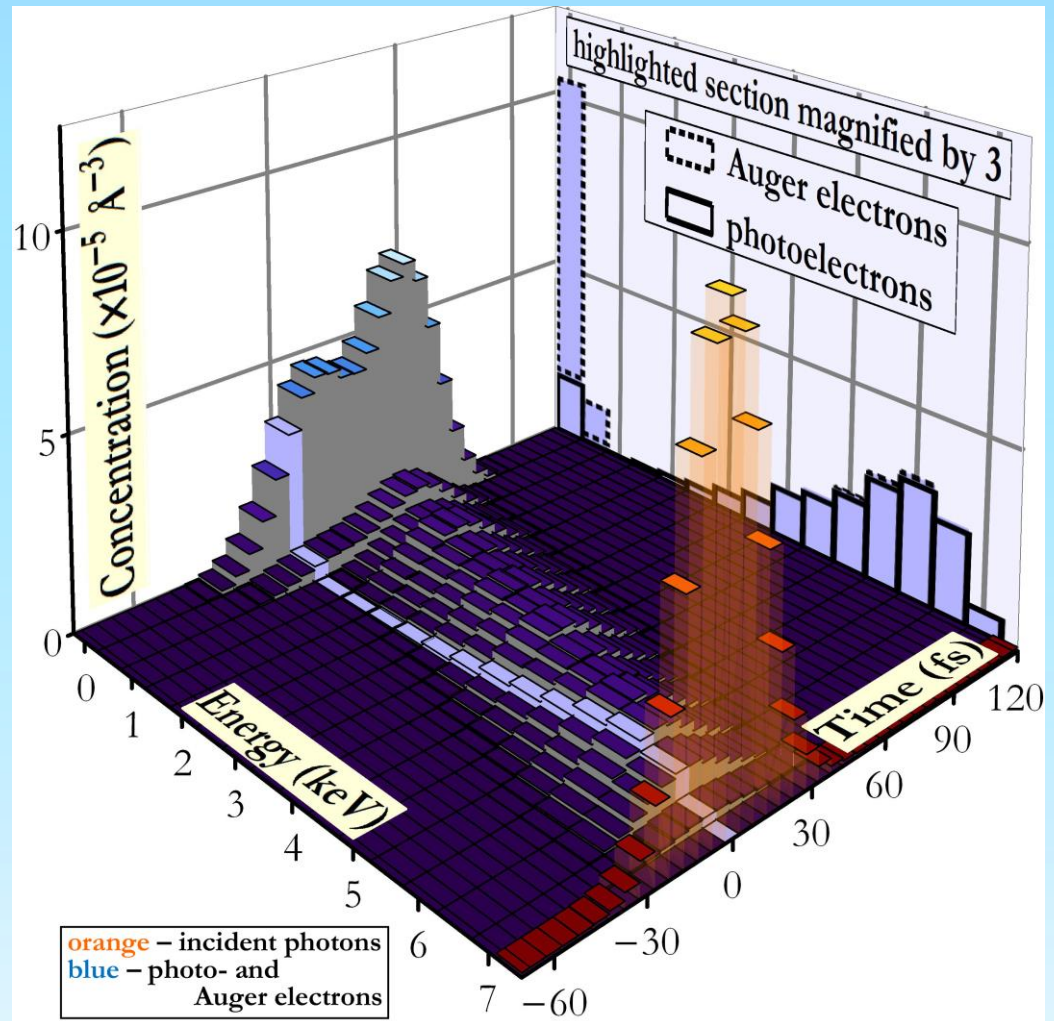
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Understanding the mechanism

Energy and time distribution of photons and electrons, Fe ionization probability

$$f_{Fe}(t) = \int \sigma_{i,e \leftrightarrow Fe}(E) \bar{v}_e(E) n_e(E, t) dE$$

$f_{Fe}(t)$ - electron-impact ionization frequency for single Fe atom

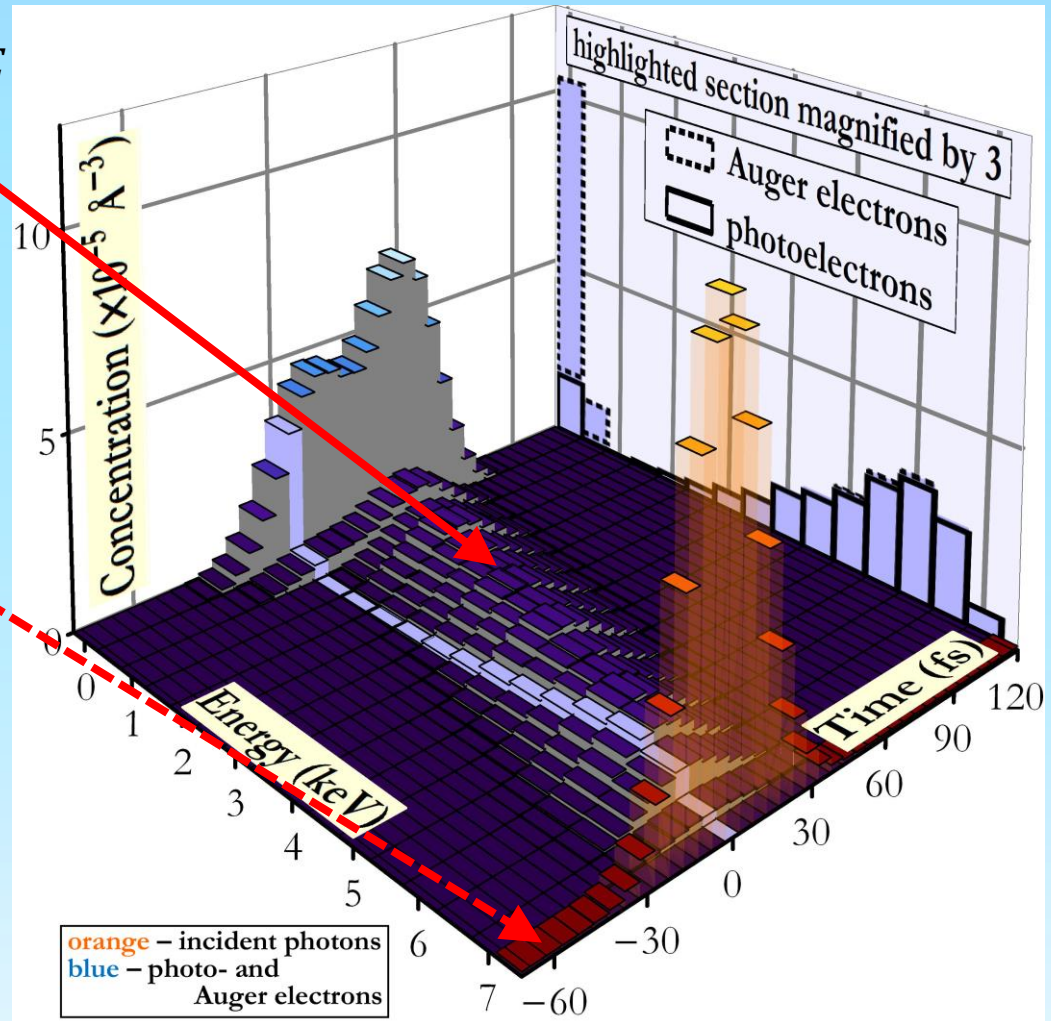
$n_e(E, t)$ - simulated energy and time evolution of the energetic electrons

$$F_{Fe}(t) = \frac{\sigma_{ph \leftrightarrow Fe}}{S} n_{ph}(t)$$

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$\sigma_{ph \leftrightarrow Fe}$ - Fe 1s photoabsorption cross section

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P. M. Banks, *Planet. Space Sci.* **14**, 1085-1103 (1966).

V. Vinogradov, V. P. Shevel'ko, *Sov. Phys. JETP* **44**, 542 (1976).

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Spectrochim. Acta, Part B **66**, 776-784 (2011).

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J. Appl. Phys. **103**, 063707 (2008).

Phys. Med. Biol. **30**, 331-335 (1985).

Understanding the mechanism

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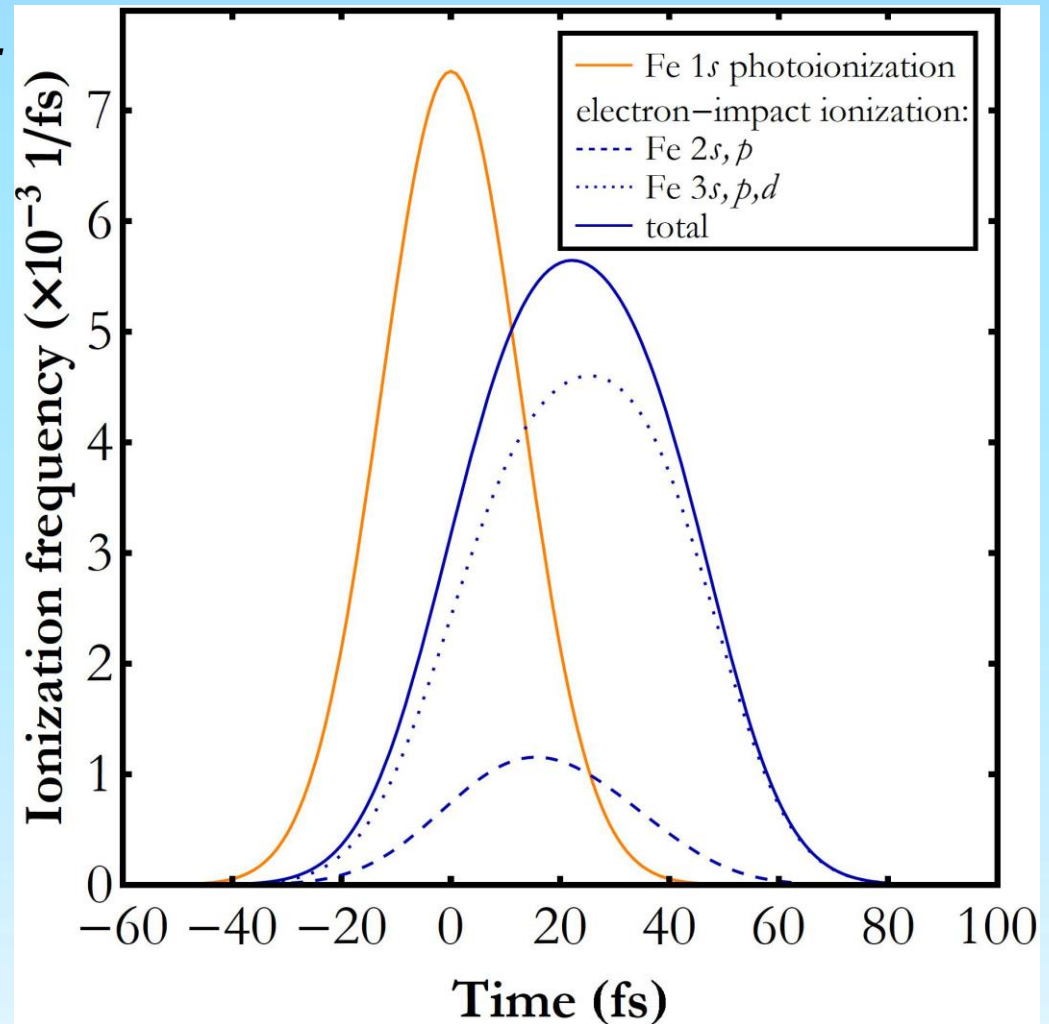
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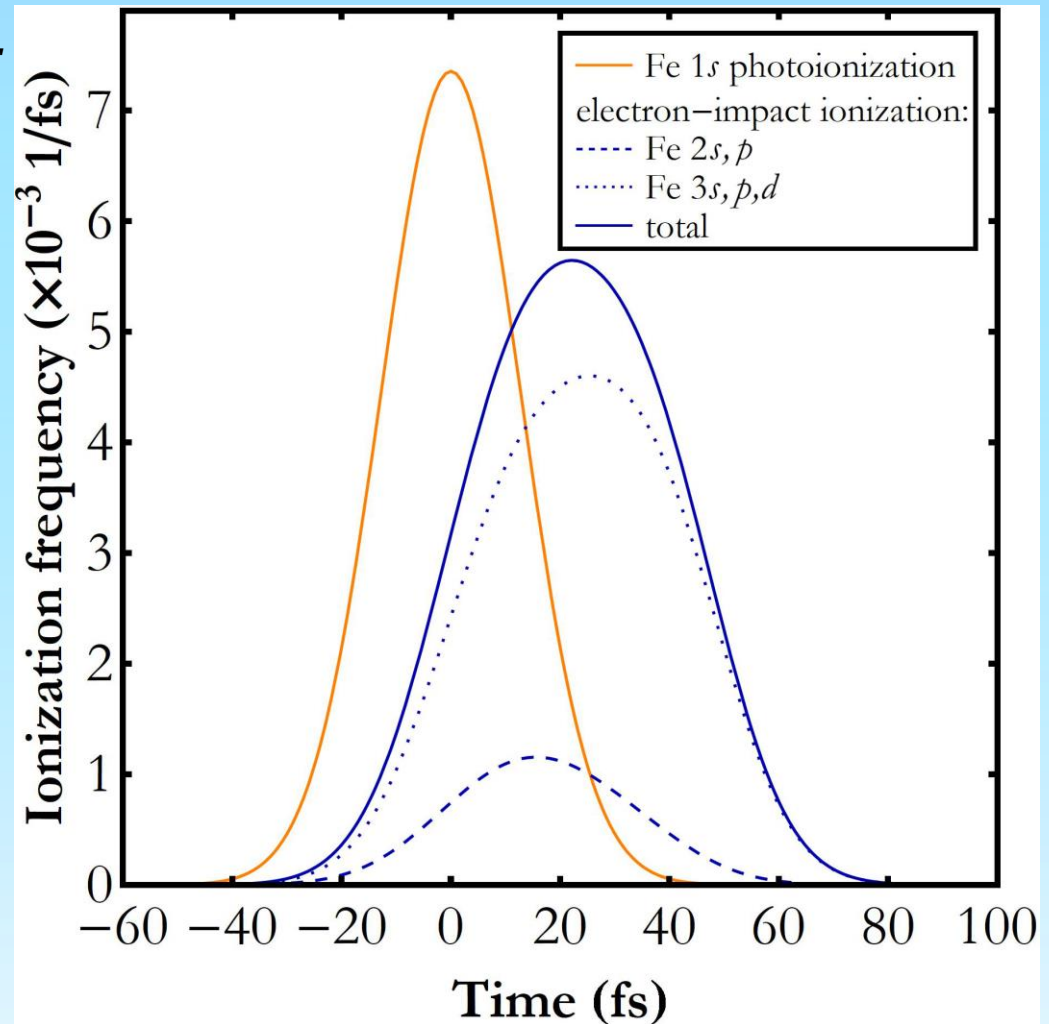
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$\sigma_{ph \leftrightarrow Fe}$ - Fe 1s photoabsorption cross section

$n_{ph}(t)$ - simulated time evolution of the incident photons

$$\frac{\int_{-\infty}^{\infty} F_{Fe}(t) \left[\int_{-\infty}^t f_{Fe}(t') dt' \right] dt}{\int_{-\infty}^{\infty} F_{Fe}(t) dt} \approx 5\%$$



P. M. Banks, *Planet. Space Sci.* **14**, 1085-1103 (1966).

V. Vinogradov, V. P. Shevel'ko, *Sov. Phys. JETP* **44**, 542 (1976).

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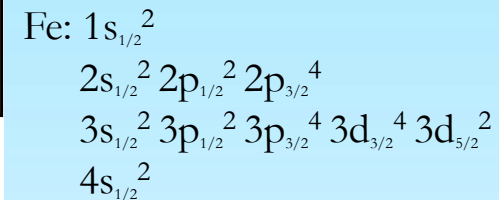
Phys. Med. Biol. **30**, 331-335 (1985).

Understanding the mechanism

Multiple ionization of Fe atoms through Auger decays – probability

Initial electron hole level	Establishment probability											
	1	2	3	4	5	6	7	8	9	10	11	12
$3d_{5/2} (M_5)$	1	0	0	0	0	0	0	0	0	0	0	0
$3d_{3/2} (M_4)$	0	1	0	0	0	0	0	0	0	0	0	0
$3p_{3/2} (M_3)$	1	0	0	0	0	0	0	0	0	0	0	0
$3p_{1/2} (M_2)$	0	0.67	0.33	0	0	0	0	0	0	0	0	0
$3s_{1/2} (M_1)$	0	0.37	0.4	0.22	0.01	0	0	0	0	0	0	0
$2p_{3/2} (L_3)$	0	0.24	0.32	0.28	0.12	0.04	0	0	0	0	0	0
$2p_{1/2} (L_2)$	0	0.18	0.27	0.27	0.16	0.08	0.03	0.01	0	0	0	0
$2s_{1/2} (L_1)$	0	0.12	0.21	0.24	0.18	0.12	0.07	0.04	0.02	0.01	0	0
$1s_{1/2} (K)$	0.02	0.15	0.21	0.22	0.15	0.1	0.06	0.04	0.02	0.02	0.01	0

Fe atoms' ionization due to only the Auger decays induced by a single electron-impact ionization reaches more than 6 electron holes on the valence levels which results in oxidation state of Fe atoms in $\text{Fe}^{2+}(\text{CN})_6$ of more than 8+.



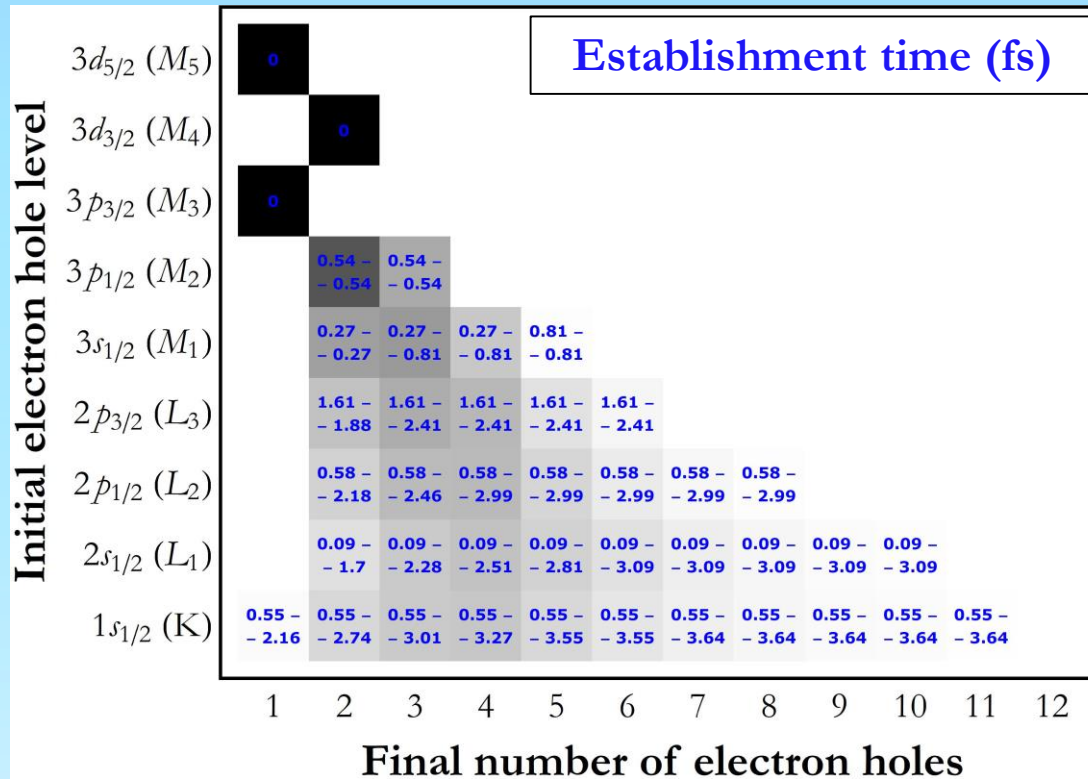
The estimated multiple electron hole-states' relative occurrences are approximately **0.29 : 0.32 : 0.21 : 0.12 : 0.04 : 0.02** for respectively **1, 2, 3, 4, 5 and 6 electron holes** or equivalently for **Fe oxidation state 3+, 4+, 5+, 6+, 7+, 8+**.

T. Schoonjans *et al.*, *Spectrochim. Acta, Part B* **66**, 776-784 (2011).

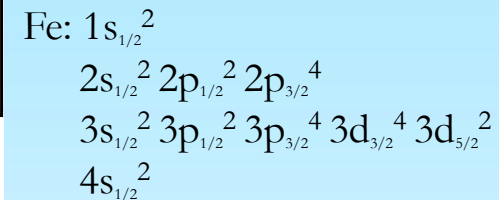
A. Thompson, X-ray Data Booklet (Lawrence Berkeley Laboratory, Berkeley, USA, ed. 3, 2009), pp. 9-64.

Understanding the mechanism

Multiple ionization of Fe atoms through Auger decays – time



Fe atoms' ionization due to only the Auger decays induced by a single electron-impact ionization reaches more than 6 electron holes on the valence levels which results in oxidation state of Fe atoms in $\text{Fe}^{2+}(\text{CN})_6$ of more than 8+.



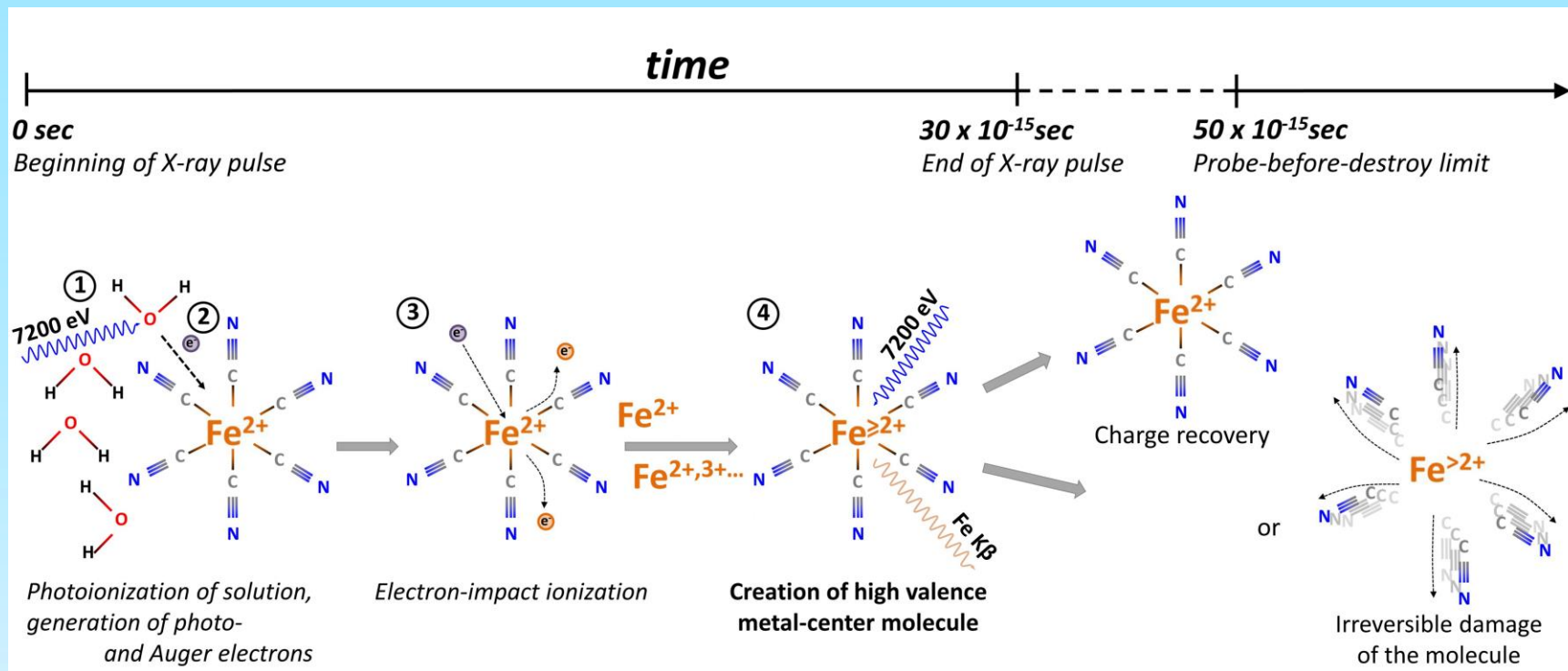
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Understanding the mechanism

Summary

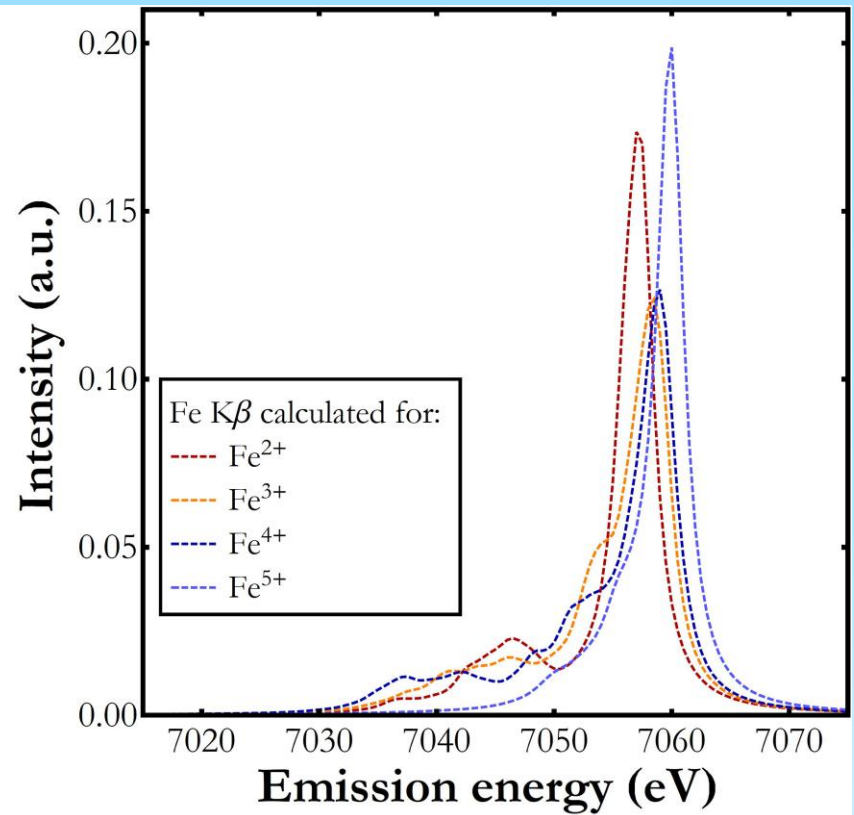
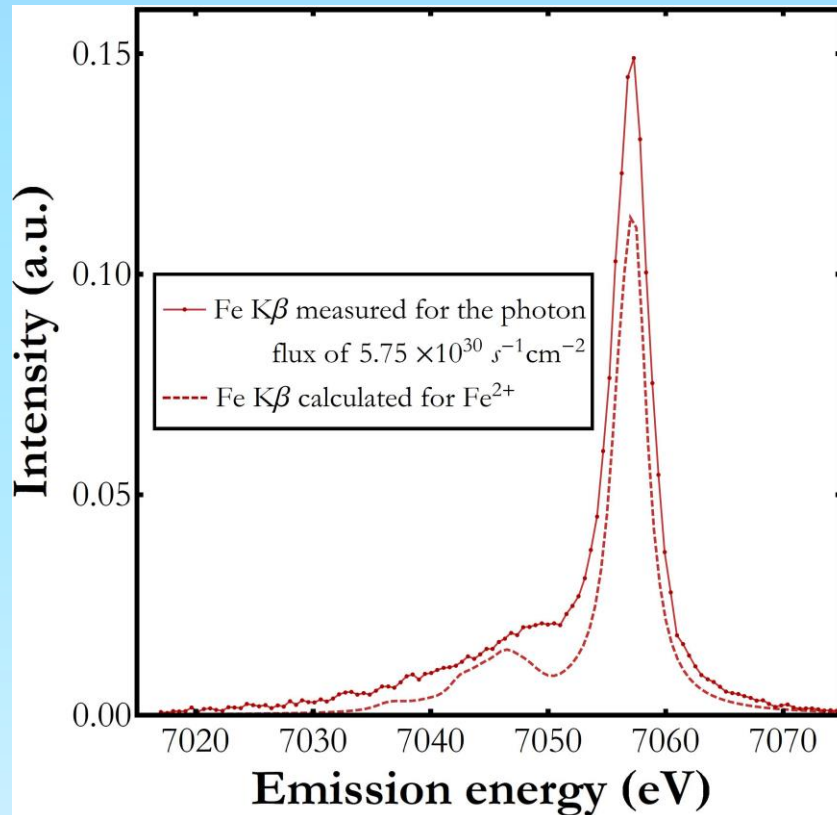


J. Kern *et al.*, *Proc. Natl. Acad. Sci. U.S.A.* **109**, 9721-9726 (2012).

S. M. Durbin, T. Clevenger, T. Graber, R. Henning, *Nat. Photonics* **6**, 111-114 (2012).

Fe K β X-ray emission spectrum calculation

Effect of Fe oxidation state in Fe(CN) $_6$

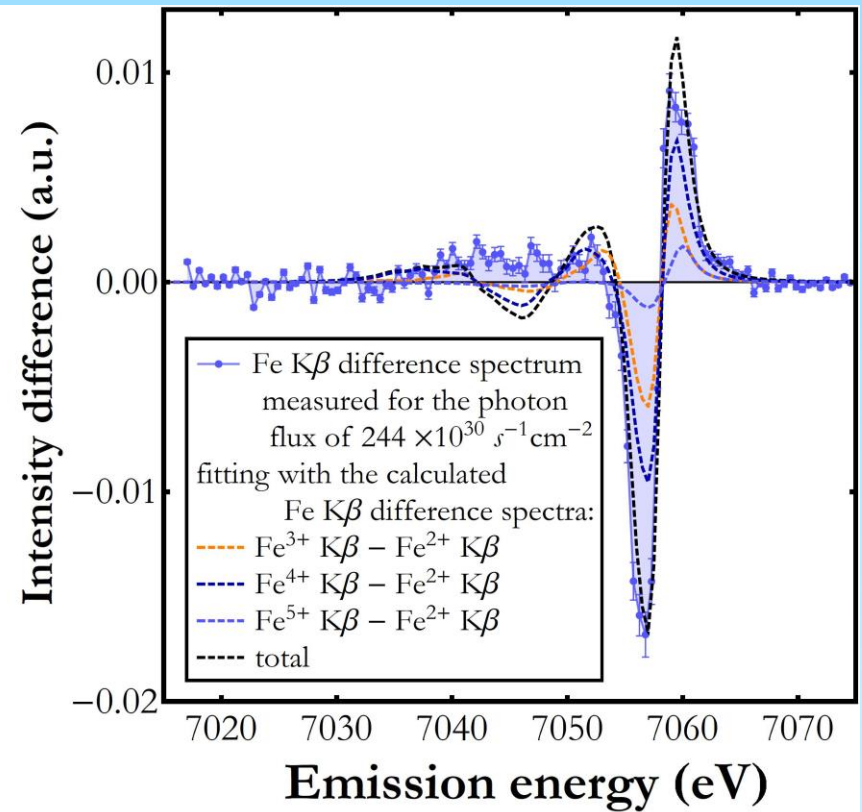
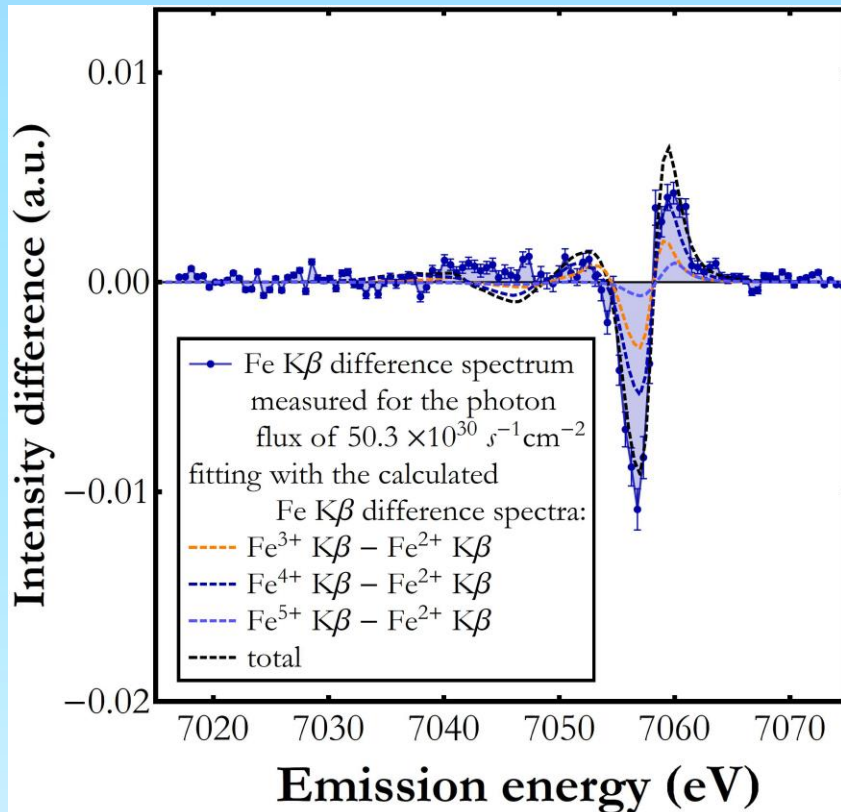


The spectra were calculated within density functional theory (DFT). The field splitting parameter $10 Dq$ of 4.2 eV was used in the multiplet calculations and metal-to-ligand charge transfer was not included.



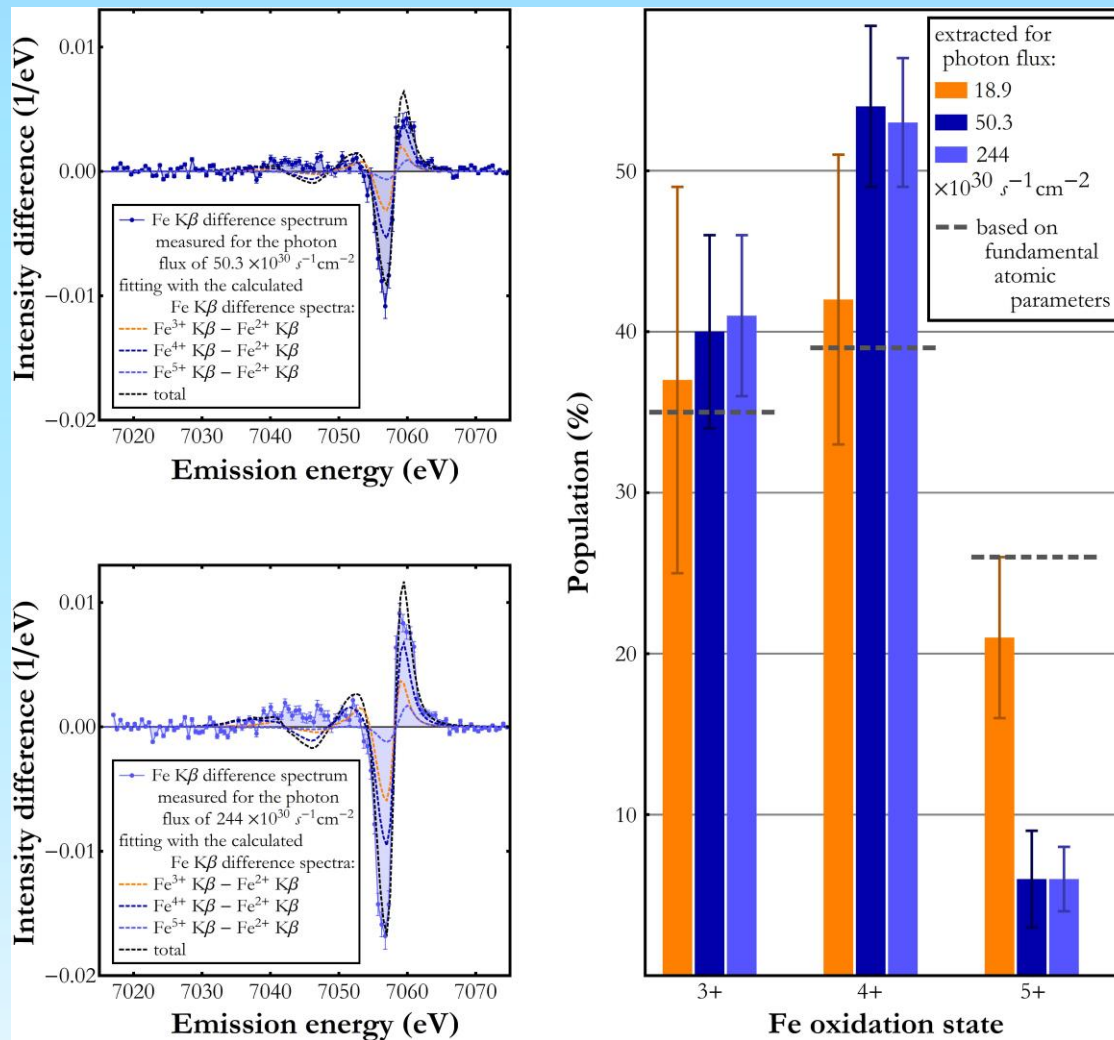
Fe $K\beta$ X-ray emission spectrum calculation

Extraction of populations of different Fe oxidation states



Fe K β X-ray emission spectrum calculation

Extraction of populations of different Fe oxidation states



Summary

- Fe K β XES for $K_4[Fe^{2+}(CN)_6 \cdot 3H_2O]/H_2O$ solution at XFEL
- 30 fs-short pulses of 4×10^{11} 7200 eV-photons at the repetition rate of 120 Hz (typical operation parameters), beam focusing with movable Be lenses
- Observed: monotonic dependence of Fe K β XES spectra on the incident XFEL beam flux
- Mechanism: multiple ionization on the molecule's Fe site caused by electron impact and Auger decays
- The molecule at this short moment (< 50 fs) is in a unique high valence state where the bonds between Fe and C atoms are broken but the molecule's constituents have not moved yet.
- The effect is solvent dependent.

Thank you for your attention.

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M. Kavčič^[5]

E. Källman^[3]

G. Knopp^[4]

M. Lundberg^[3]

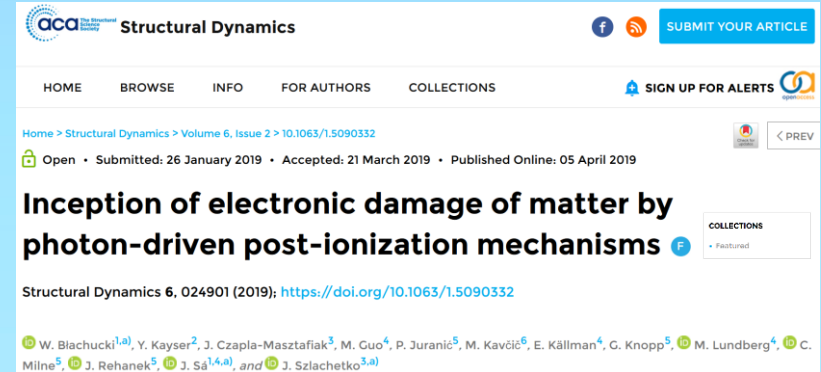
C. Milne^[4]

J. Rehanek^[4]

J. Sá^[3,6]

J. Szlachetko^[1]

W. Blachucki *et al.*, *Structural Dynamics* 6, 024901 (2019).



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